

REGIOSPECIFIC SYNTHESIS OF  $\alpha\beta$ -UNSATURATED EPOXIDES (1)

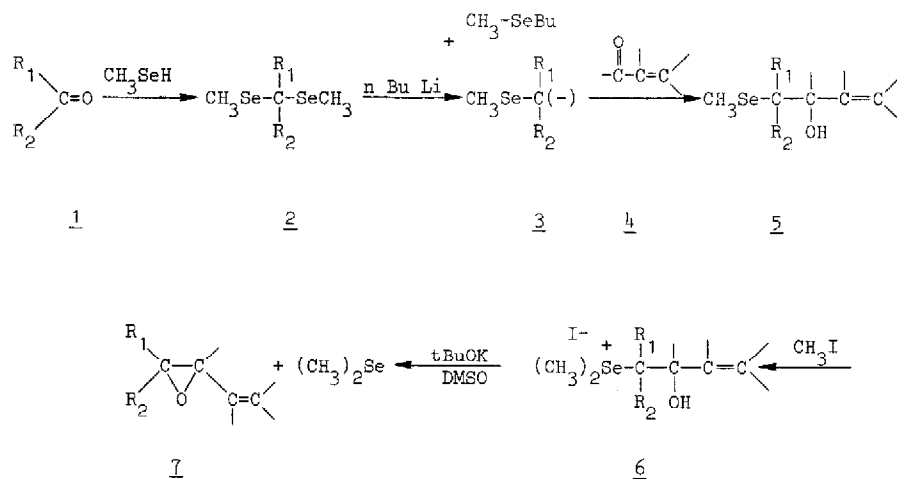
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Until now the most powerful [C-C] connective (2) route to an  $\alpha\beta$ -unsaturated epoxide was the alkylidene transfer from a sulfur ylid (3) to an  $\alpha\beta$ -unsaturated aldehyde or ketone. This method is of wide application only in the methylene case; there is no report of mono alkylidene transfer and it was shown that isopropylidene diphenyl sulfurane reacts specifically on the carbon-carbon double bond of cyclohexenone (4), leading to a cyclopropane, instead of an epoxide.

Recently we described (5,6) a new general synthon for [C-C] connective synthesis of epoxides, using  $\alpha$ -seleno carbanions 3 as one of the two building blocks. We report here an extension of our previous work (5) for the regiospecific synthesis of  $\alpha\beta$ -unsaturated epoxides 1 (Scheme I)

SCHEME I



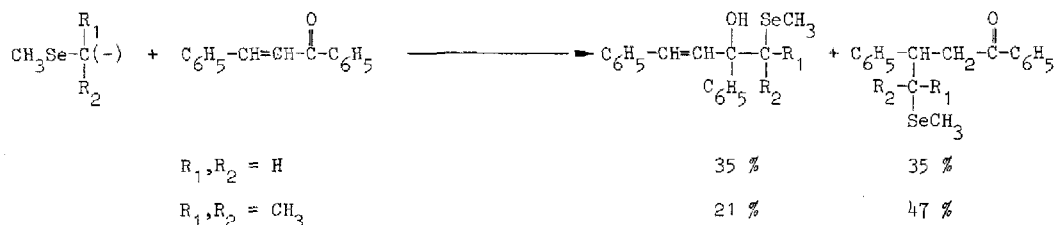
This synthon, which allows the synthesis even of tetrasubstituted epoxides, takes advantage of the easy formation of  $\alpha$ -seleno carbanions from carbonyl compounds 1 [1  $\rightarrow$  2, selenol, H<sup>+</sup>, ~ 80 % yield] [2  $\rightarrow$  3 n-butyllithium, THF - 78°C, 1h] and their high propensity to react on the

carbon oxygen bond of an  $\alpha\beta$ -unsaturated aldehyde or ketone [THF,  $-78^\circ\text{C}$ , 2h;  $-25^\circ\text{C}$ , 1h] leading specifically to an  $\alpha\beta$ -unsaturated  $\alpha'$ -seleno alcohol 5 in good yield (Table I).

Seleno alcohol 5 is easily separated from by-products (7) and in some cases, eg. 7g, the separation of the two diastereoisomers is readily performed by preparative thick layer chromatography on silicagel (PLC). The seleno alcohol 5 is quantitatively transformed (8) to the corresponding selenonium salt 6 [ $\text{CH}_3\text{I}$ , 2 ~ 4h,  $25^\circ\text{C}$ ], which is in turn transformed to the desired epoxide 7 in good yield [tBuOK in DMSO,  $25^\circ\text{C}$ , 1h]. In our hands these epoxides are quite labile and they are destroyed on silicagel PLC even when buffered (pH 7); they are generally obtained in good analytical purity without any purification. The dimethyl selenide formed during this reaction is highly volatile.

Of the different carbonyl compounds used, chalcone is the only case where  $\alpha$ -seleno carbanions react to a large extent on the carbon-carbon double bond. This tendency of the 1-4 (9,10) versus the 1-2 (3b,3c) addition is well documented (Scheme II).

SCHEME II



$\alpha$ -phenyl seleno carbanions react as in the methyl seleno case, leading to the corresponding  $\beta$ -hydroxy phenyl selenides (11), but as already mentioned (5) in the case of saturated carbonyl compounds, we were unable to synthesize  $\beta$ -hydroxy selenonium salts when  $\text{R}_1, \text{R}_2 = \text{alkyl}$ , and rearranged products are formed (12) (Scheme III).

SCHEME III

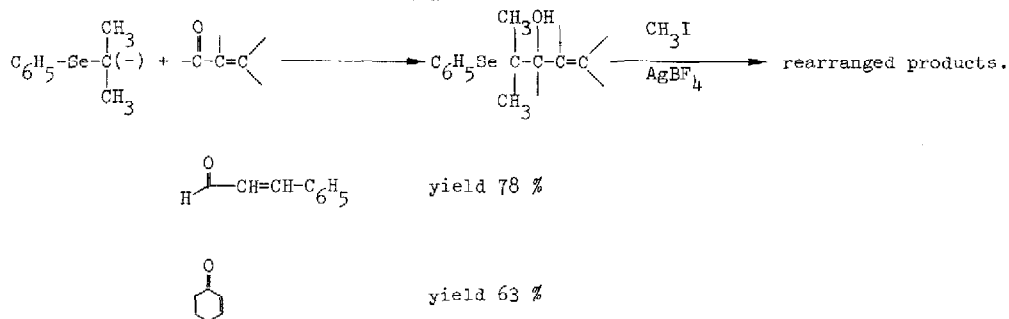





TABLE I

According to Scheme I

R <sub>1</sub>	R <sub>2</sub>	<u>4</u>	<u>5</u> * 13	<u>7</u> *** 13	
H	H	$\text{C}_6\text{H}_5-\text{CH}=\text{CH}-\overset{\text{O}}{\underset{\text{O}}{\text{C}}}-\text{CH}_3$	87	75	a
CH <sub>3</sub>	CH <sub>3</sub>	$\text{C}_6\text{H}_5-\text{CH}=\text{CH}-\overset{\text{O}}{\underset{\text{O}}{\text{C}}}-\text{CH}_3$	80	89	b
H	CH <sub>3</sub>	$\text{C}_6\text{H}_5-\text{CH}=\text{CH}-\overset{\text{O}}{\underset{\text{O}}{\text{C}}}-\text{H}$	90	94	c
H	CH <sub>3</sub>		65	98	d
CH <sub>3</sub>	CH <sub>3</sub>		45	75	e
CH <sub>3</sub>	CH <sub>3</sub>		55	95	f
H	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>5</sub>	$\text{CH}_3-\text{CH}=\text{CH}-\overset{\text{O}}{\underset{\text{O}}{\text{C}}}-\text{CH}_3$	65	40(80)	g
CH <sub>3</sub>	CH <sub>3</sub>	$\text{CH}_3-\text{CH}=\text{CH}-\overset{\text{O}}{\underset{\text{O}}{\text{C}}}-\text{CH}_3$	45	50	h

\* Yield of purified product

\*\*\* Yield of analytically pure product (salt 6 washed with ether, without any purification at the epoxide stage), based on 5.() Based on reacted 5.

*References*

- 1) A preliminary report of this work was presented by the authors at the 25th I.U.P.A.C. Congress held in Jerusalem (Israel) Juli 1975
- 2) For definition of this term see E.J. Corey, *Pure and Appl. Chemistry* 14, 19 (1967)
- 3a) A.W. Johnson, *Ylid Chemistry*, Academic Press, New York (1966)
- 3b) E.J. Corey and M. Chaykovsky, *J. Amer. Chem. Soc.*, 87, 1353 (1965)
- 3c) For selenium ylid case : W. Dumont, P. Bayet and A. Krief, *Angew. Chem., Int. Ed.* 13, 309 (1974)
- 4a) E.J. Corey and M. Jautelat, *J. Amer. Chem. Soc.*, 89, 3913 (1967)
- 4b) For ylids derived from sulfoximines see C.R. Johnson, R.A. Kirchhoff, R.J. Reischer and G.F. Katekar, *J. Amer. Chem. Soc.* 95, 4287 (1973)
- 5) W. Dumont and A. Krief, *Angew. Chem. Int. Ed.* 14, 350 (1975)
- 6) D. Van Ende, W. Dumont and A. Krief, *Angew. Chem. Int. Ed.* 14, 700 (1975)
- 7) Methyl butyl selenide is easily removed by evaporation (0.1 mm Hg, 25°C)
- 8) The salt is washed with ether and sometimes some unreacted starting material is separated
- 9) A. Merz and G. Märkl, *Angew. Chem. Int. Ed.* 12, 845 (1973)
- 10) M.S. Kharasch and O. Reinmuth, *Grignard Reactions of Nonmetallic Substances*, Prentice-Hall, (1954)
- 11)  $\alpha\beta$ -unsaturated  $\alpha'$ seleno alcohols 5 are easily transformed to  $\alpha\beta, \alpha'\beta'$ -unsaturated alcohols in high yield : unpublished results from our laboratory
- 12) Work is in progress to determine the structure of these compounds
- 13) Spectroscopic data (IR, NMR, Mass Spectra) are in accord with the proposed structures.